Some New Analogues of Pethidine.* Part I.†

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Substituted ethyl 1-alkyl-4-phenylpiperidine-4-carboxylates in which the alkyl group carries a nitrogenous heterocyclic residue have been prepared from ethyl 4-phenylpiperidine-4-carboxylate for pharmacological test as analgesics.

The discovery of the analgesic properties of pethidine (I) was followed by the synthesis and pharmacological evaluation of a large number of related substances. Pethidine, although not very potent on a weight basis, has held its place as the most widely used synthetic analgesic because of its low toxicity. In the present work, further analogues of pethidine have been synthesised in the hope that more potent substances retaining the low toxicity of pethidine might be obtained.

$$(I) \begin{array}{c} Ph \\ EtO_2C \end{array} \begin{array}{c} NMe \\ EtO_2C \end{array} \begin{array}{c} Ph \\ N\cdot Y\cdot X \quad 2HCI \end{array} \hspace{0.5cm} (II)$$

The substances synthesised have the general formula (II), in which Y represents an alkyl group carrying a heterocyclic residue (X) (cf. Table); the two heterocyclic systems in each case are linked through their nitrogen atoms by a two (or three) carbon-atom chain (Y). Although two derivatives of pethidine carrying a basic substituent at the nitrogen atom have been reported, none has been described hitherto in which the 1-alkyl group was substituted by a heterocyclic residue.

The starting material for the syntheses reported here was ethyl 4-phenylpiperidine-4carboxylate,² an intermediate in one of the commercial methods³ of making pethidine. One of two methods was used for the preparation of the desired substituted substances: preferably the secondary base (I; H in place of Me) was treated with the desired substituted alkyl halide, but in the cases where difficulties might have arisen (Nos. 4, 7, 8, and 9 in Table) the

3 B.I.O.S. Final Report No. 766, p. 60 et seq.

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† Submitted in honour of the seventieth birthday of Sir Ian Heilbron, D.S.O., F.R.S.

Schaumann, Arch. Path. Pharmakol., 1940, 196, 127.
 Eisleb, Ber., 1941, 74, 1433; cf. Walton, B.P. 592,016.

secondary base was first alkylated with ethylene chlorohydrin; the resulting 1-hydroxyethyl compound was converted into the chloroethyl compound which reacted readily with the heterocyclic bases chosen. The desired products were isolated and characterised as hydrochlorides.

The substances prepared were submitted to pharmacological tests, the results of which have been reported.4 When this work was in progress (1954-55), it was generally accepted 5 that replacement of the N-methyl group of pethidine by other groups decreased or abolished analgesic activity. It was surprising therefore, that four of the substances, namely, Nos. 3, 5, 10, and 11 in the Table, had very considerable analgesic potency; ethyl

		Pethid	ine an	alogue	s (II).					
No.	Y	X					M	M. p. (decomp.)		
1	CH ₂ ·CH ₂	Pyrrolidino						259—262°		
	сн. сн.	Piperidi						270-	272	
3	сн, сн,	Morpholino						264-266		
4	CH₂·CH₂	Piperazino						244-246		
5	CH ₂ ·CH ₂	Tetrahydro-1: 4-thiazin-4-yl 266								
2 3 4 5 6 7 8	CH ₂ ·CH ₂	4-Ethoxycarbonyl-4-phenylpiperidino 241—24								
7	CH ₂ ·CH ₂	4-Methylpiperidino 270—272								
8	CH ₂ ·CH ₂	1:2:3:6-Tetrahydropyridino 257—259								
9	CH ₂ ·CH ₃	4-Hydro	xypipe	ridino				250—		
10	CH, CH, CH,	Morphol						24		
11	CHMe·CH ₂	Morpholino 235—237								
			Found	(%)			Requir	ed (%)		
No.	Formula	c	Found	(%) N	Cl	c	Requir H	ed (%)	cì	
No.	C ₉₀ H ₉₉ O ₉ N ₂ Cl ₉	C 59·0			Cl 17·65					
1	C ₉₀ H ₉₉ O ₉ N ₂ Cl ₉	59·0 59·7	H 8·0 7·6	N 7·15 6·75	$\substack{17.65\\17.2}$	C 59·5 60·4	H 7·9 8·2	N 6·95 6·7	Cl 17·6 17·0	
1	C ₂₀ H ₃₂ O ₂ N ₂ Cl ₂ C ₂₀ H ₃₄ O ₂ N ₂ Cl ₂ C ₂₀ H ₃₄ O ₂ N ₂ Cl ₂	59·0 59·7 56·9	H 8·0 7·6 7·65	N 7·15	17.65	C 59·5 60·4 57·3	H 7·9 8·2 7·7	N 6-95	Cl 17·6	
1	C ₂₀ H ₃₂ O ₂ N ₂ Cl ₂ C ₂₀ H ₃₄ O ₂ N ₂ Cl ₂ C ₂₀ H ₃₂ O ₃ N ₂ Cl ₂ C ₂₀ H ₃₂ O ₃ N ₂ Cl ₂ C ₃₀ H ₃₄ O ₂ N ₂ Cl ₃	59·0 59·7 56·9 53·0	H 8·0 7·6 7·65 7·6	N 7·15 6·75 6·65	$\substack{17.65\\17.2}$	C 59.5 60.4 57.3 52.9	H 7·9 8·2 7·7 7·5	N 6·95 6·7 6·7	Cl 17·6 17·0	
1 2 3 4 5	C ₂₀ H ₃₂ O ₂ N ₂ Cl ₂ C ₂₀ H ₃₄ O ₂ N ₃ Cl ₂ C ₂₀ H ₃₅ O ₃ N ₂ Cl ₂ C ₂₀ H ₃₄ O ₂ N ₃ Cl ₃ C ₂₀ H ₃₂ O ₂ N ₃ SCl ₃	59·0 59·7 56·9 53·0 55·6	H 8·0 7·6 7·65 7·6 7·55	N 7·15 6·75 6·65 — 6·35	17·65 17·2 16·9	C 59·5 60·4 57·3 52·9 55·15	H 7·9 8·2 7·7 7·5 7·4	N 6·95 6·7 6·7 6·45	Cl 17·6 17·0 16·9	
1 2 3 4 5	C ₂₀ H ₃₂ O ₂ N ₂ Cl ₂ C ₂₀ H ₃₄ O ₂ N ₃ Cl ₂ C ₂₀ H ₃₂ O ₃ N ₃ Cl ₂ C ₂₀ H ₃₄ O ₂ N ₃ Cl ₃ C ₂₀ H ₃₂ O ₂ N ₂ SCl ₃ C ₃₀ H ₄₂ O ₄ N ₂ Cl ₂	59·0 59·7 56·9 53·0 55·6 62·9	H 8·0 7·6 7·65 7·6 7·55 7·35	N 7·15 6·75 6·65 — 6·35 4·75	$\substack{17.65\\17.2}$	C 59.5 60.4 57.3 52.9 55.15 63.7	H 7·9 8·2 7·7 7·5 7·4 7·4	N 6·95 6·7 6·7 6·45 4·95	Cl 17·6 17·0	
1 2 3 4 5	C ₂₀ H ₃₉ O ₂ N ₂ Cl ₂ C ₂₀ H ₃₄ O ₂ N ₂ Cl ₂ C ₂₀ H ₃₄ O ₂ N ₂ Cl ₂ C ₂₀ H ₃₅ O ₃ N ₂ Cl ₂ C ₂₀ H ₃₀ O ₂ N ₂ Cl ₂ C ₂₀ H ₃₂ O ₂ N ₂ SCl ₂ C ₃₀ H ₄₂ O ₄ N ₃ Cl ₂ C ₂₂ H ₄₅ O ₃ N ₂ Cl ₂	59.0 59.7 56.9 53.0 55.6 62.9 61.25	H 8·0 7·6 7·65 7·65 7·55 7·35 7·9	N 7·15 6·75 6·65 — 6·35 4·75 6·3	17·65 17·2 16·9	C 59.5 60.4 57.3 52.9 55.15 63.7 61.2	H 7·9 8·2 7·7 7·5 7·4 7·4 8·4	N 6.95 6.7 6.7 	Cl 17·6 17·0 16·9	
1 2 3 4 5	$\begin{array}{c} C_{20}H_{33}O_2N_2Cl_2\\ C_{20}H_{34}O_2N_2Cl_2\\ C_{20}H_{34}O_2N_2Cl_2\\ C_{20}H_{34}O_2N_2Cl_2\\ C_{20}H_{32}O_2N_2Cl_3\\ C_{20}H_{32}O_2N_2SCl_2\\ C_{30}H_{42}O_4N_2Cl_2\\ C_{22}H_{36}O_2N_2Cl_3\\ C_{21}H_{32}O_2N_2Cl_2\\ \end{array}$	59.0 59.7 56.9 53.0 55.6 62.9 61.25 59.9	H 8·0 7·6 7·65 7·6 7·55 7·35 7·9 7·35	N 7·15 6·75 6·65 — 6·35 4·75 6·3 6·75	17·65 17·2 16·9 — — 12·5 —	C 59·5 60·4 57·3 52·9 55·15 63·7 61·2 60·7	H 7·9 8·2 7·7 7·5 7·4 7·4 8·4 7·7	N 6·95 6·7 6·7 	Cl 17·6 17·0 16·9	
1 2 3 4 5 6 7 8 9	C ₈₀ H ₃₈ O ₃ N ₂ Cl ₂ C ₂₀ H ₃₄ O ₃ N ₂ Cl ₂ C ₂₀ H ₃₄ O ₃ N ₂ Cl ₂ C ₂₀ H ₃₄ O ₃ N ₂ Cl ₃ C ₂₀ H ₃₂ O ₂ N ₂ SCl ₂ C ₃₀ H ₄₂ O ₄ N ₃ Cl ₂ C ₂₂ H ₃₆ O ₃ N ₂ Cl ₂ C ₂₁ H ₃₂ O ₂ N ₂ Cl ₂ C ₂₁ H ₃₂ O ₃ N ₂ Cl ₂ C ₃₁ H ₃₄ O ₃ N ₂ Cl ₂	59.0 59.7 56.9 53.0 55.6 62.9 61.25 59.9 58.0	H 8·0 7·6 7·65 7·55 7·35 7·35 7·35 7·8	N 7·15 6·75 6·65 — 6·35 4·75 6·3 6·75 6·15	17·65 17·2 16·9 — 12·5 —	C 59·5 60·4 57·3 52·9 55·15 63·7 61·2 60·7 58·2	H 7·9 8·2 7·7 7·5 7·4 7·4 8·4 7·7	N 6·95 6·7 6·7 	Cl 17·6 17·0 16·9 — — 12·55 —	
1 2 3 4 5	$\begin{array}{c} C_{20}H_{33}O_2N_2Cl_2\\ C_{20}H_{34}O_2N_2Cl_2\\ C_{20}H_{34}O_2N_2Cl_2\\ C_{20}H_{34}O_2N_2Cl_2\\ C_{20}H_{32}O_2N_2Cl_3\\ C_{20}H_{32}O_2N_2SCl_2\\ C_{30}H_{42}O_4N_2Cl_2\\ C_{22}H_{36}O_2N_2Cl_3\\ C_{21}H_{32}O_2N_2Cl_2\\ \end{array}$	59.0 59.7 56.9 53.0 55.6 62.9 61.25 59.9	H 8·0 7·6 7·65 7·6 7·55 7·35 7·9 7·35	N 7·15 6·75 6·65 — 6·35 4·75 6·3 6·75	17·65 17·2 16·9 — — 12·5 —	C 59·5 60·4 57·3 52·9 55·15 63·7 61·2 60·7	H 7·9 8·2 7·7 7·5 7·4 7·4 8·4 7·7	N 6·95 6·7 6·7 	Cl 17·6 17·0 16·9	

1-2'-morpholinoethyl-4-phenylpiperidine-4-carboxylate, in particular, was 3-7 times as potent as pethidine. These appeared to be the first known cases of potent analgesics carrying a large substituent at the "piperidine" nitrogen, though one further closely related example, phenethylnorpethidine, has lately been reported.⁶

It was particularly striking that the four highly active pethidine analogues of this series contained the morpholine or tetrahydro-1:4-thiazine ring system. This led to the synthesis, by simple new routes (cf. Experimental section), of two known morpholine derivatives, 1:2-dimorpholinoethane and 2-morpholinoethylcyclohexane, and of the new N-(2-4'.-hydroxypiperidinoethyl)morpholine; these substances were devoid of analgesic activity. The conclusion was reached, therefore, that whilst analgesic activity did not reside in the morpholine moiety of the molecule itself, the morpholine ring system had an important bearing on analgesic potency in this series of pethidine analogues. Further work is in progress.

EXPERIMENTAL

Intermediates.—The intermediate alkyl halides were prepared by recorded methods. 2-Pyrrolidinoethyl chloride hydrochloride had m. p. 172—173° (Wright et al. give m. p. 173.5— 174°); 2-piperidinoethyl chloride hydrochloride had m. p. 228° (Dunlop 8 gives m. p. 231°, but Marckwald and Frobenius 9 give m. p. 208°); 2-morpholinoethyl chloride hydrochloride had

- Millar and Stephenson, Brit. J. Pharmacol., 1956, 11, 27; Green and Ward, ibid., p. 32.
 Cf. Burger, "Medicinal Chemistry," p. 174 Interscience, New York, 1951; Braenden, Eddy, and Halbach, Bull. Wld. Hlth. Org., 1955, 13, 937.
 Perrine and Eddy, J. Org. Chem., 1956, 21, 125.
 Wright, Kolloff, and Hunter, J. Amer. Chem. Soc., 1948, 70, 3098.

 - Dunlop, J., 1912, 101, 1998.
 Marckwald and Frobenius, Ber., 1901, 34, 3556.

m. p. 182° (Mason and Block 10 give m. p. 182-182.5°); 2-(tetrahydro-1: 4-thiazin-1-yl)ethyl chloride hydrochloride had m. p. 208° (Gilman and Woods 11 give m. p. 206-208°); 3-morpholinopropyl chloride hydrochloride had m. p. 168° (Adams and Whitmore 12 give m. p. 168-170°); and 1-methyl-2-morpholinoethyl chloride hydrochloride had m. p. 176-177° (Attenburrow et al.13 give m. p. 176-178°).

Ethyl 1-2'-Chloroethyl-4-phenylpiperidine-4-carboxylate (I; R = •CH,•CH,•Cl).—Ethyl 4phenylpiperidine-4-carboxylate (20 g.), m. p. ca. 33°, b. p. 150°/1·8 mm. (cf. Eisleb 2) [hydrobromide, m. p. 165—166° (Found: N, 4.55; Br, 25.4. C₁₄H₂₀O₂NBr requires N, 4.45; Br, 25.45%)], and ethylene chlorohydrin (6 ml.) were refluxed in ethyl acetate for 2.5 hr. For isolation of the intermediate (II; $YX = {}^{\bullet}CH_2{}^{\bullet}CH_2{}^{\bullet}OH$), the ethyl acetate was removed and the liquid remaining distilled. Ethyl 1-2'-hydroxyethyl-4-phenylpiperidine-4-carboxylate (mentioned, but not described by Schaumann 1) has b. p. $125-128^{\circ}/1-1.5$ mm., $n_{\rm p}^{19}$ 1.5215 (Found: C, 70·1; H, 8·6; N, 4·75. Calc. for $C_{16}H_{23}O_3N$: C, 69·3; H, 8·35; N, 5·05 $\frac{5}{N}$).

To prepare the product (II; YX = CH₂CH₂Cl), the stirred cool mixture was slowly treated with thionyl chloride (8 ml.) and, when addition was complete, the solution was boiled for a short time until crystallisation started. The hydrochloride of 1-2'-chloroethyl-4-phenylpiperidine-4carboxylate was collected, and recrystallised from aqueous alcohol. When pure it (ca. 8-5 g.) had m. p. 219—220° (decomp.) (Found: C, 58.4; H, 6.9; N, 4.05; Cl, 21.4. C₁₆H₂₂O₂NCl,HCl requires C, 57.8; H, 6.9; N, 4.2; Cl, 21.4%).

Preparation of Tertiary Bases.—(A) Ethyl 1-2'-morpholinoethyl-4-phenylpiperidine-4-carboxylate. Ethyl 4-phenylpiperidine-4-carboxylate (7 g.), 2-morpholinoethyl chloride (4.5 g.), and anhydrous sodium carbonate (1.8 g.) were boiled in alcohol (50 ml.) for several hours. The suspension was filtered and the filtrate concentrated under reduced pressure. Ethanolic hydrogen chloride was then added to precipitate the crude product; recrystallisation from aqueous ethanol gave the pure dihydrochloride (cf. Table) (9.8 g.), m. p. 264-266° (decomp.). Ethyl 1-2'-morpholinoethyl-4-phenylpiperidine-4-carboxylate, regenerated from the dihydrochloride, had b. p. $188-192^{\circ}/0.5$ mm., $n_{\rm D}^{18}$ 1.5276 (Found: C, 69.4; H, 8.85; N, 8.45. $C_{20}H_{30}O_3N_2$ requires C, 69.3; H, 8.75; N, 8.1%), and gave a picrate, m. p. $247-248^{\circ}$ (decomp.) (Found: N, 14.25. $C_{20}H_{30}O_3N_2, 2C_6H_3O_7N_3$ requires N, 13.9%).

(B) Ethyl 1-2'-piperazinoethyl-4-phenylpiperidine-4-carboxylate. Ethyl 1-2'-chloroethyl-4phenylpiperidine-4-carboxylate hydrochloride (5.5 g.) (see above) in alcohol was neutralised with alcoholic sodium ethoxide (1 equiv.); piperazine (3.6 g.) was added and the mixture boiled for 4 hr. After concentration under reduced pressure, addition of alcoholic hydrogen chloride gave the desired trihydrochloride, which was recrystallised from aqueous alcohol and then had m. p. 244—246° (decomp.).

1: 2-Dimorpholinoethane.--Morpholine (10 g.) and 2-morpholinoethyl chloride (17 g.) were boiled for 1 hr.; two layers were formed. To the mixture was added an excess of aqueous ammonia, and the organic bases were extracted into chloroform. Vacuum-distillation gave, besides unchanged morpholinoethyl chloride (4.5 g.), b. p. 90°/12 mm., the desired 1:2-dimorpholinoethane (12 g.), b. p. 140-150°/12 mm., m. p. 70-72° (Knorr and Brownsden 14 give m. p. 74°, b. p. 153—154°/9 mm.).

2'-Morpholinoethylcyclohexane.—The decarboxylated condensation product (76 g.), b. p. 97— 99°/15 mm. (cf. Cope et al. 15), of cyanoacetic acid and cyclohexanone on hydrogenation over Raney nickel in alcoholic ammonia at 55—60°/100 atm. for 24 hr. gave 2'-aminoethylcyclohexane (63 g.), b. p. $70^{\circ}/15$ mm., $188^{\circ}/1$ atm., $n_{\rm p}^{19}$ 1·4720 (Wallach 18 gives b. p. $188-189^{\circ}$, $n_{\rm p}^{19}$ 1·4647). This amine (30 g.) and di-(2-chloroethyl) ether (34 g.) in boiling ethanol (100 ml.) over sodium carbonate (25 g.) for 8 hr. yielded, besides much unchanged starting material, 2'-morpholinoethylcyclohexane (12 g.), b. p. 130°/12 mm., which gave a hydrochloride, m. p. 253—254° (Blicke and Zienty ¹⁷ give b. p. 132—134°/12 mm.; hydrochloride, m. p. 260—261°).

4-(2-4'-Hydroxypiperidinoethyl)morpholine.—4-Hydroxypiperidine (24.7 g.), b. p. 110— 125°/10—15 mm. (Bowden and Green 18 give b. p. 110—115°/10 mm.), and 2-morpholinoethyl chloride (37 g.) were boiled in alcohol (150 ml.) over sodium carbonate (13 g.) for several hours.

<sup>Mason and Block, J. Amer. Chem. Soc., 1940, 62, 1443.
Gilman and Woods, ibid., 1945, 67, 1843.
Adams and Whitmore, ibid., p. 735.
Attenburrow, Elks, Hems, and Speyer, J., 1949, 510.
Knorr and Brownsden, Ber., 1902, 35, 4472.
Cope, D'Addieco, Whyte, and Glickman, Org. Synth., 1951, 31, 25.
Wallach, Annalen, 1907, 353, 284; 1908, 359, 311.
Blicke and Zienty, J. Amer. Chem. Soc., 1939, 61, 771.
Bowden and Green. J. 1952, 1164.</sup>

¹⁸ Bowden and Green, J., 1952, 1164.

The suspension was filtered, the filtrate concentrated, and 20% alcoholic hydrogen chloride added: 4-(2-4'-hydroxypiperidinoethyl)morpholine dihydrochloride separated; on crystallisation from aqueous alcohol it (50 g.) had m. p. 295° (decomp.) (Found: C, 46·35; H, 8·45; N, 9·55; Cl, 24·3. $C_{11}H_{22}O_2N_2$,2HCl requires C, 46·0; H, 8·4; N, 9·75; Cl, 24·7%). Regeneration with aqueous alkali, extraction into chloroform, and distillation gave the base, b. p. 155—158°/2—3 mm., which crystallised rapidly on cooling and, after recrystallisation from ether-light petroleum, had m. p. 65—66° (Found: C, 61·6; H, 10·1; N, 13·1. $C_{11}H_{22}O_2N_2$ requires C, 61·65; H, 10·35; and N, 13·05%).

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